on your test

<u>Differences Between Pure Substance & Mixtures</u>

Are NOT easy to observe. Scientist took hundreds of thousands of years of investigating to figure out if pure substance existed

- Must investigate the properties of material to find out that they are always the same. If they are the exact same in all parts of the material, no matter what part of the world the material comes from, then you say you have a pure substance.

It is difficult for scientist to determine if a substance is pure because

they need to check it and compare it to others all over the world.

This is hard because the world is HUGE.



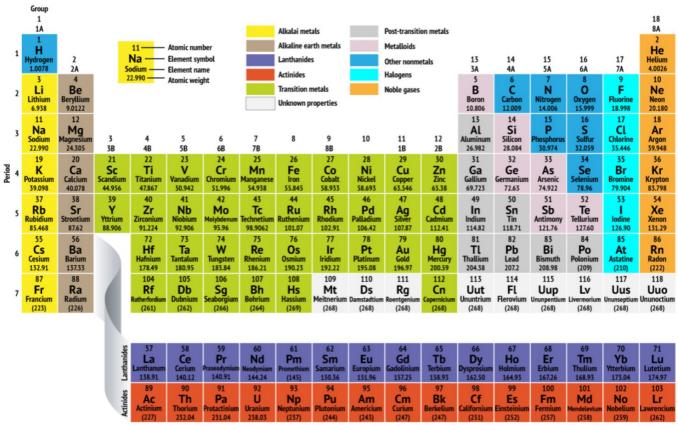
This ring is gold and if you turned it over, you would expect to see the same propertied on the reverse- the same colour, texture, and shininess.



Found naturally on earth

www.LiveScience.com

Periodic Table of the Elements



SOURCES: National Institute of Standards and Technology, International Union of Pure and Applied Chemistry

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We <u>can</u> form solutions by mixing one or more materials together (like sugar and water). This is dissolving.

Not all mixtures form solutions.

What other things get dissolved?

The Attraction of Particles when Dissolving can be explained by the Particle Theory

Particles stay together because

i) they are attracted to each other. Sometimes particles can become attracted to other particles (like sugar and water).

ii)Particles are always moving.

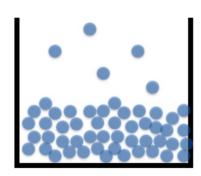
https://phet.colorado.edu/en/simulation/legacy/soluble-salts

Molecules of water and salt combining - advanced

Why'd the Ocean Stop Getting Saltier? - YouTube
Why is the Dead Sea so salty? - YouTube

The Dead Sea Is DYING - YouTube

The Benefits of Dead Sea Products on The Doctors TV Show - YouTube

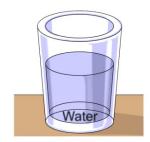


A drop of water of water evaporates into gas.

The particles are always moving, however, and some are always on the outside of the drop. These outside particles occasionally jump off into the air. Overtime, all the particles jump off. They still exist, but they are independent and free to move about as gas.



Why Do Some Materials NOT Dissolve?



Solvent

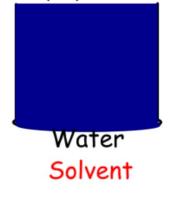


There is usually less solute than solvent in a solution (more solvent than solute.)

Solute - is the substance that dissolves in a solvent to form a solution (ex. the salt in water)

Solvent - is the substance that dissolves a solute to form a solution (Ex. usually water)

<u>Solution - is a</u> mixture of 2 or more materials that come together to form one set of properties that blend together



Sugar Cube Solute

